



MAXIMIL

Maximum Operation Temperature (MOT) - Bestimmung von
Applikationsdaten ionischer Flüssigkeiten (IL) mit Hilfe
nicht-isothermer Messungen zur thermischen Stabilität

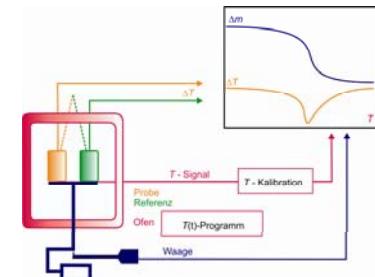
IL BASED SYNTHESIS



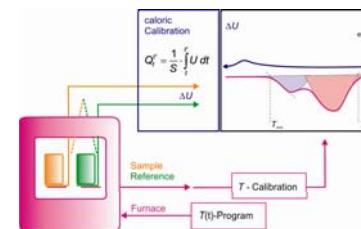
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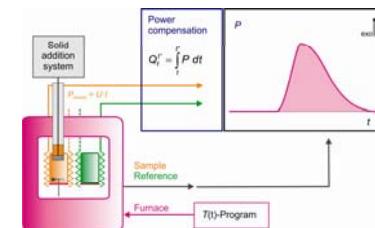
Simultaneous Thermal Analysis (DTA/TG/MS)



Dynamic Calorimetry (DSC)



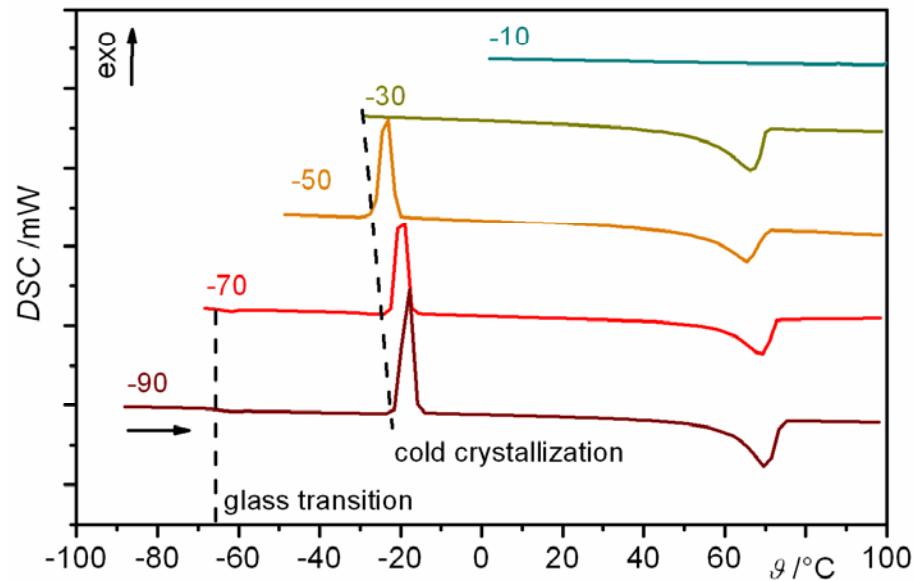
Reaction Calorimetry



Modeling

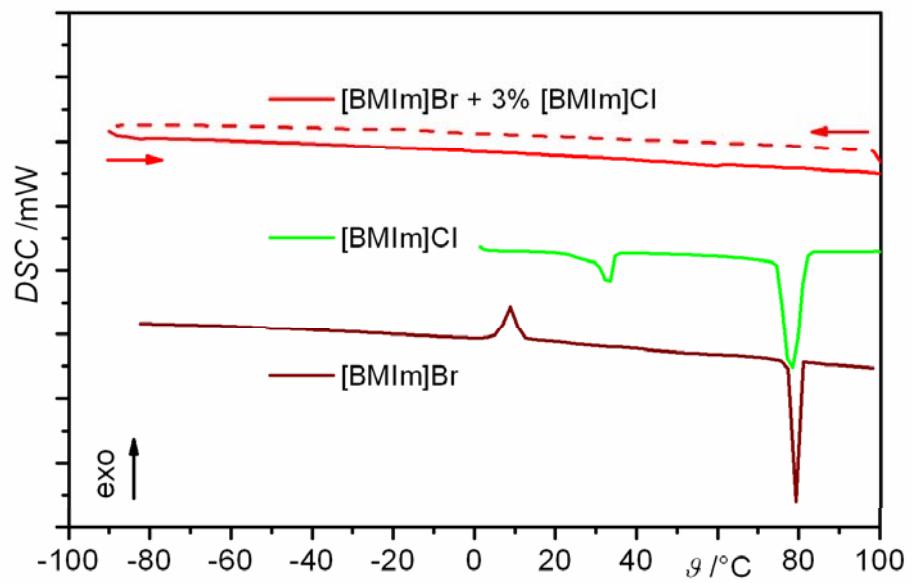
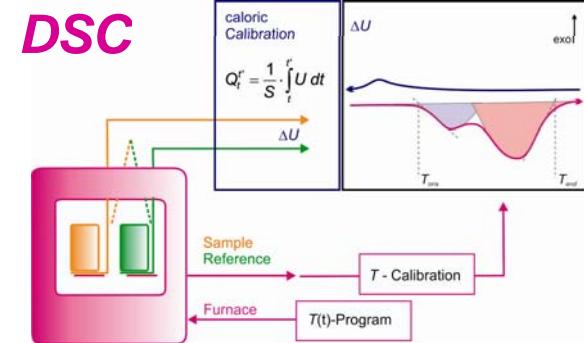


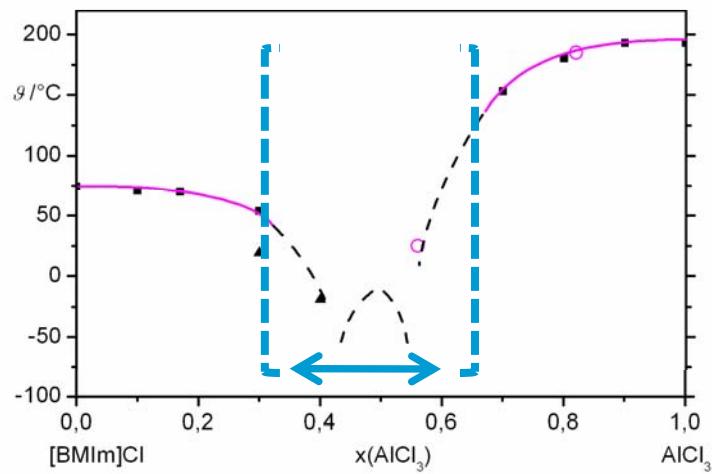
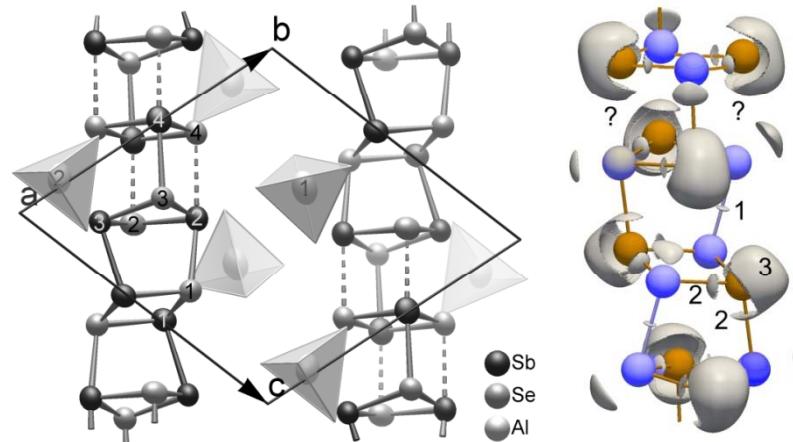
SUPERCOOLING AND GLASS FORMATION



[BMIm] X ($X = \text{Cl}, \text{Br}$)

Thermochim. Acta 2013, 573, 162.
Thermochim. Acta 2015, 604, 129.



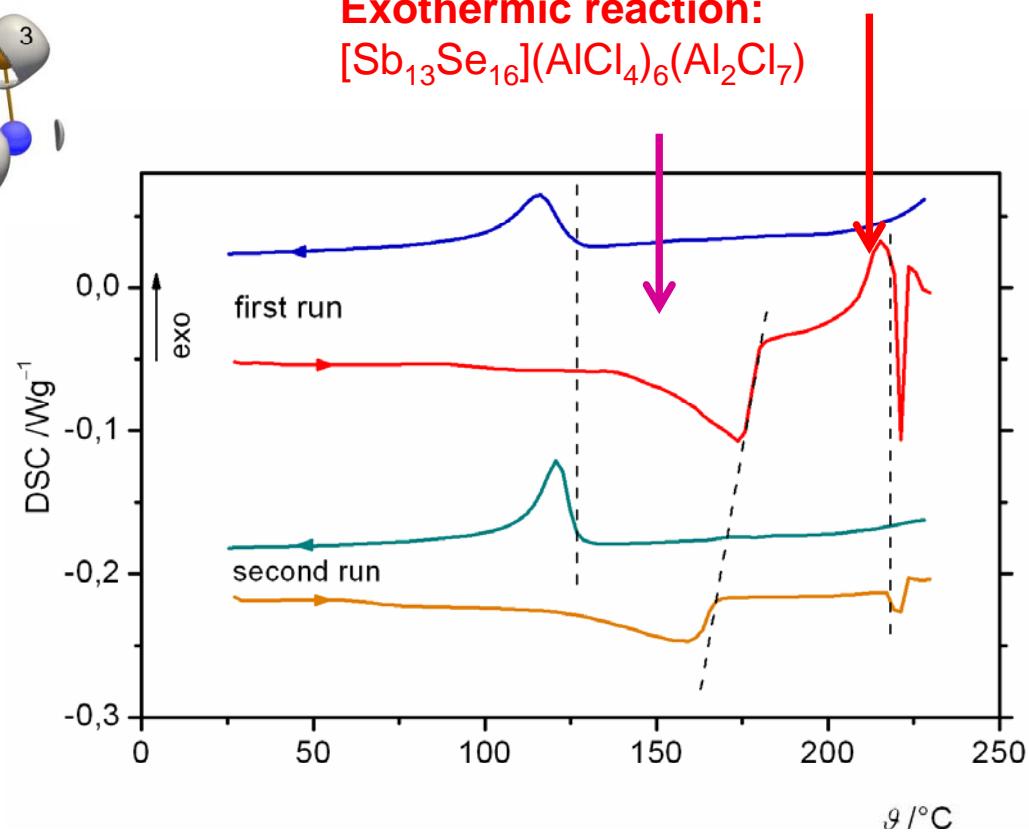


Endothermic melting:



Formation of [Sb₂Se₂]AlCl₄

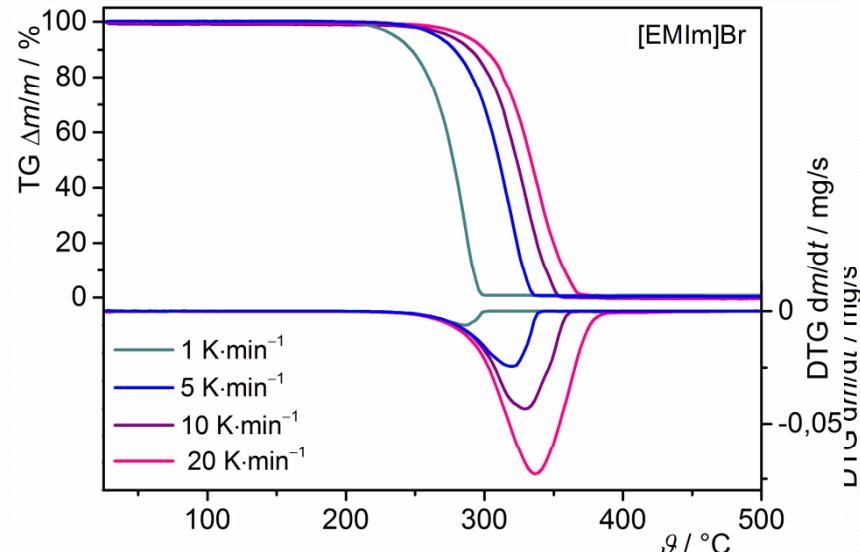
Exothermic reaction:



THERMAL STABILITY OF IONIC LIQUIDS

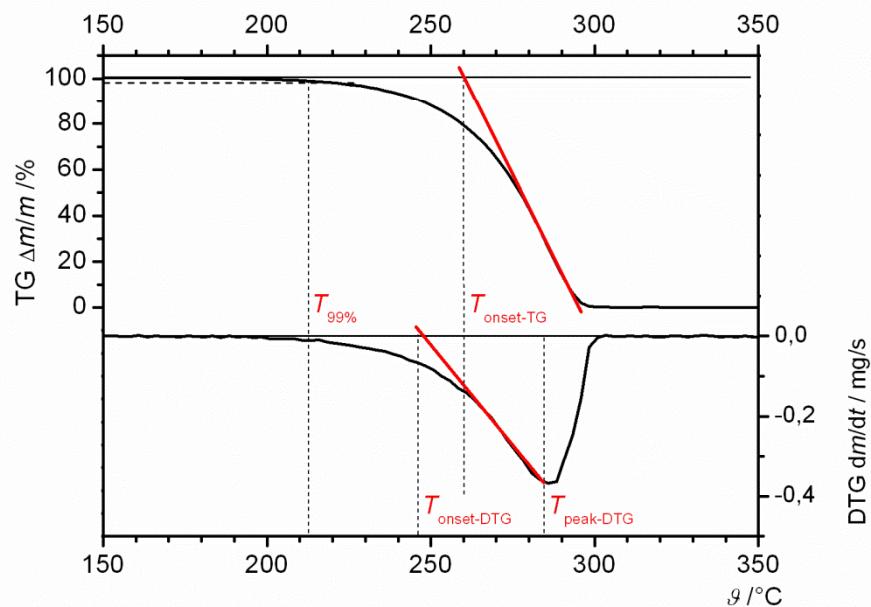
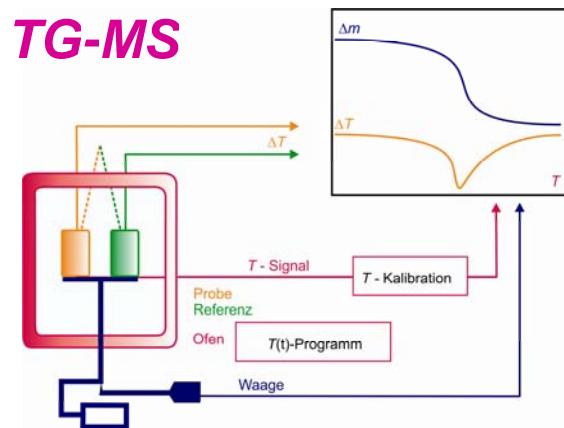


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How to evaluate the „real“ temperature of thermal decomposition?

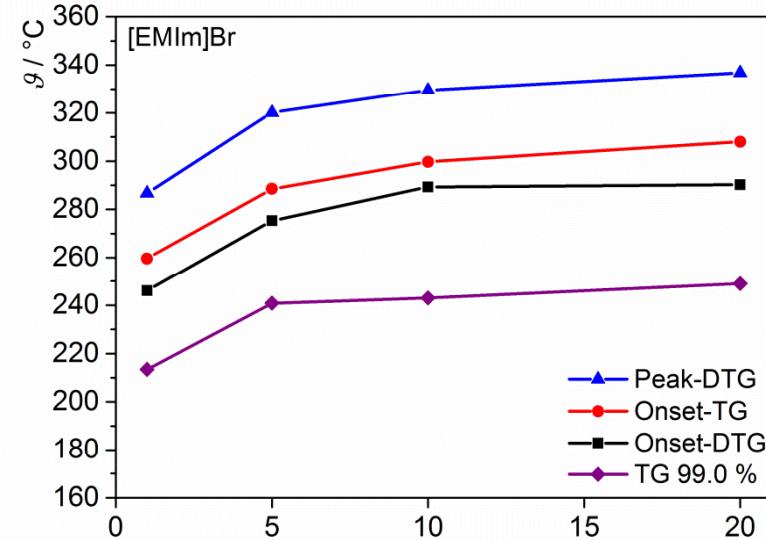
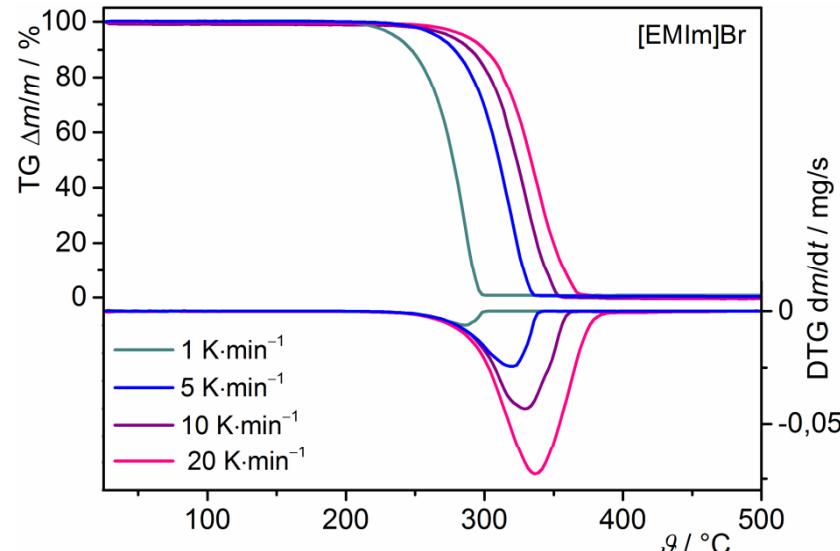
TG-MS



Thermal Stability of Ionic Liquids



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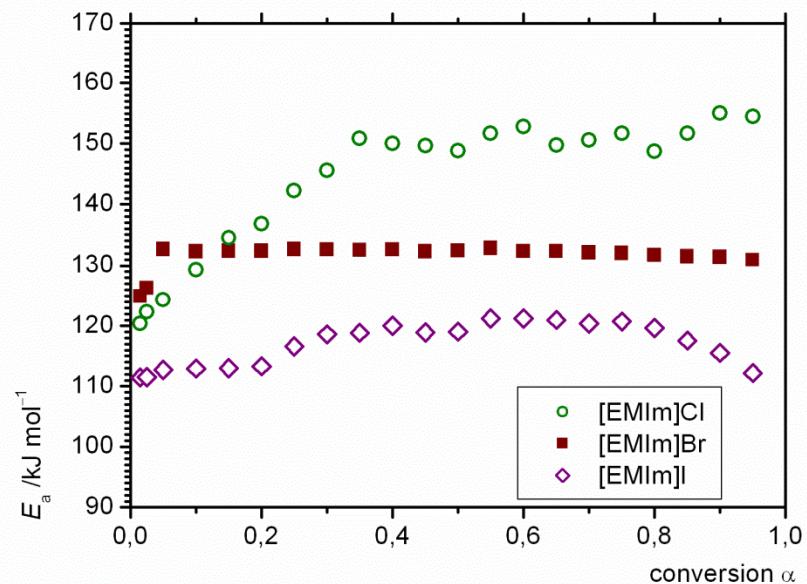


Integral isoconversional method:

heating rate β
vs. temperature T at α

$$\ln\left(\frac{\beta_i}{T_{\alpha,i}^2}\right) = A - \frac{E_a}{R \cdot T_\alpha}$$

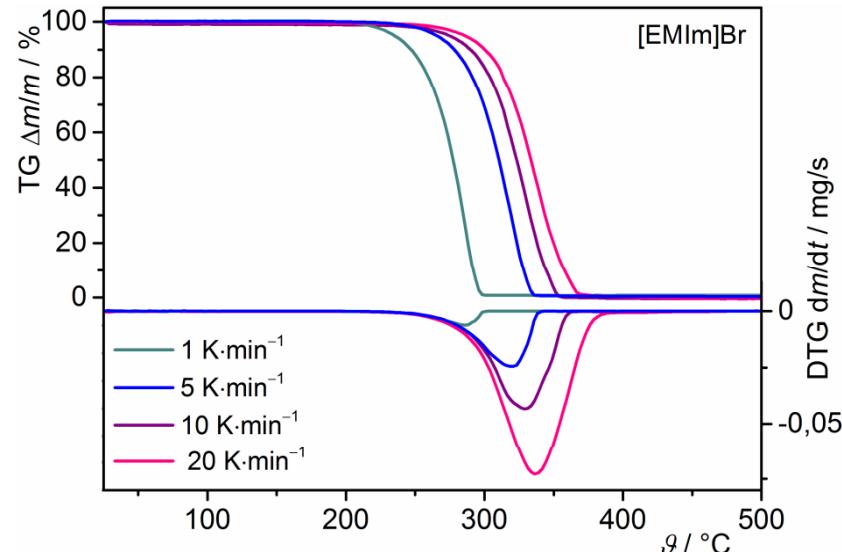
Thermochim. Acta 2015, 604, 129.



THERMAL STABILITY OF IONIC LIQUIDS

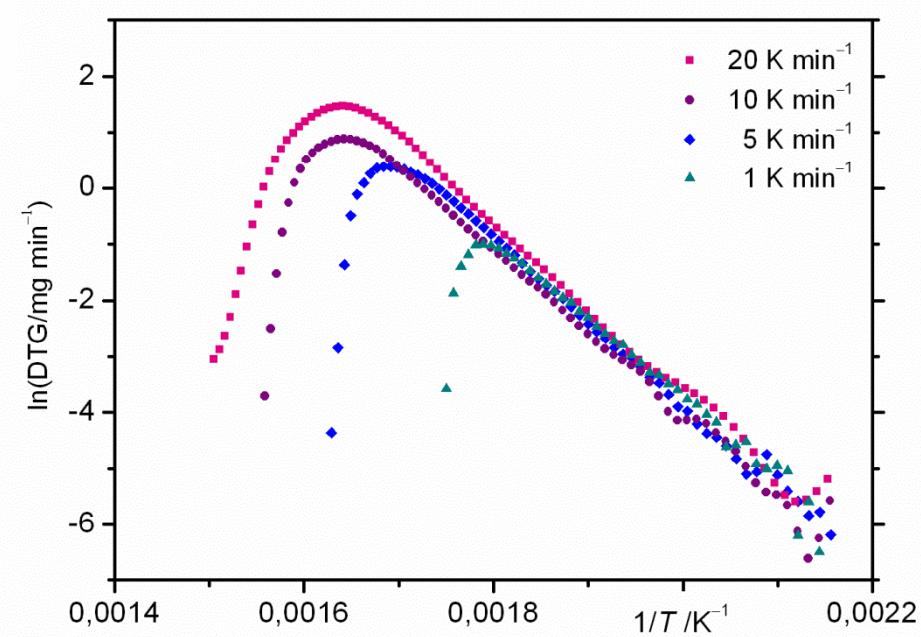


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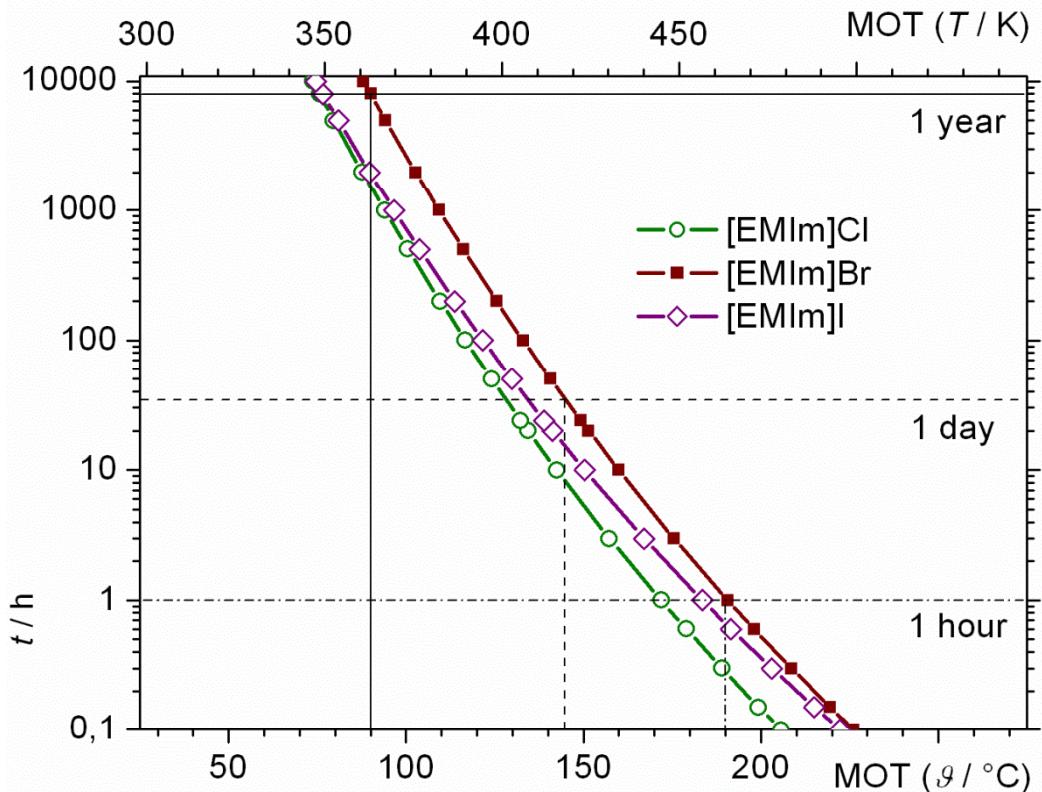
Thermal Decomposition Kinetics

$$\ln\left(-\frac{dm}{dt}\right) = \ln(k_0) - \frac{E_a}{R} \cdot \frac{1}{T} + \ln(m)$$



MAXIMUM OPERATION TEMPERATURE

$$T_{max}(MOT) = \frac{E_a}{R \cdot [4.6 + \ln(k_0) \cdot t_{max}]}, \quad (\alpha = 1\%)$$



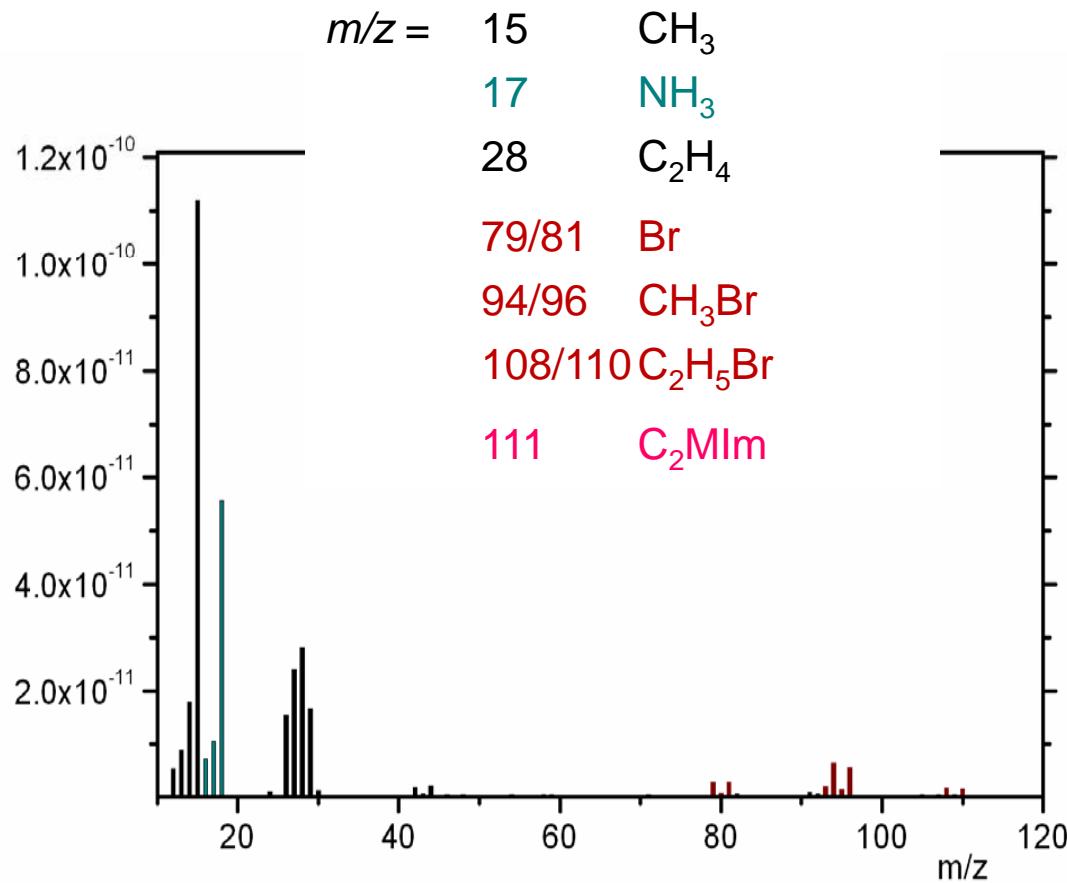
Seeberger, Andresen, Jess,
Phys. Chem. Chem. Phys. **2009** 11 9375.

THERMAL DECOMPOSITION MECHANISM



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[EMIm]Br

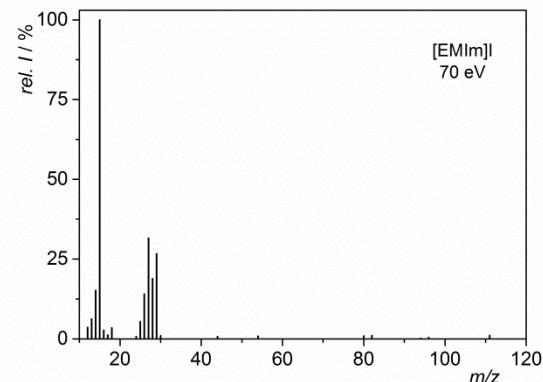
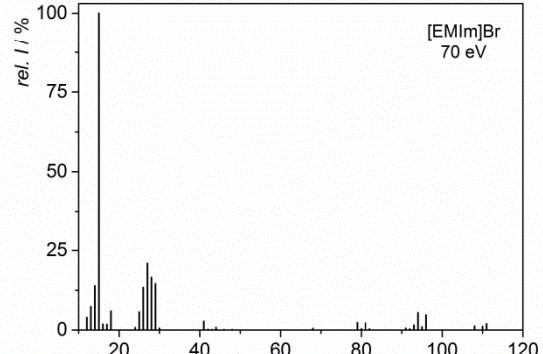
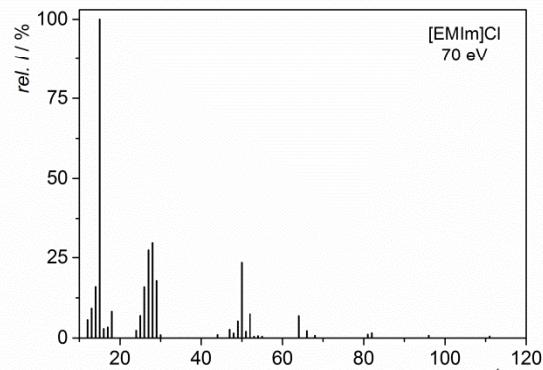


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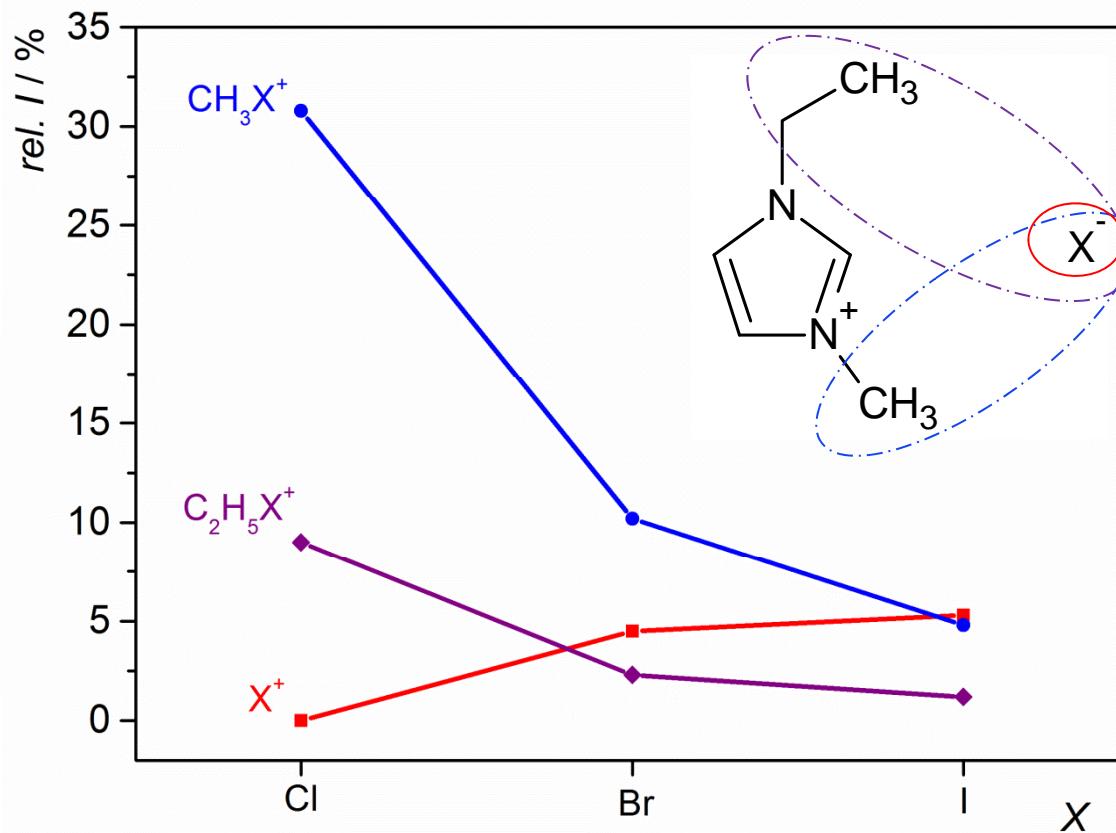
THERMAL DECOMPOSITION MECHANISM



=> Halide anion

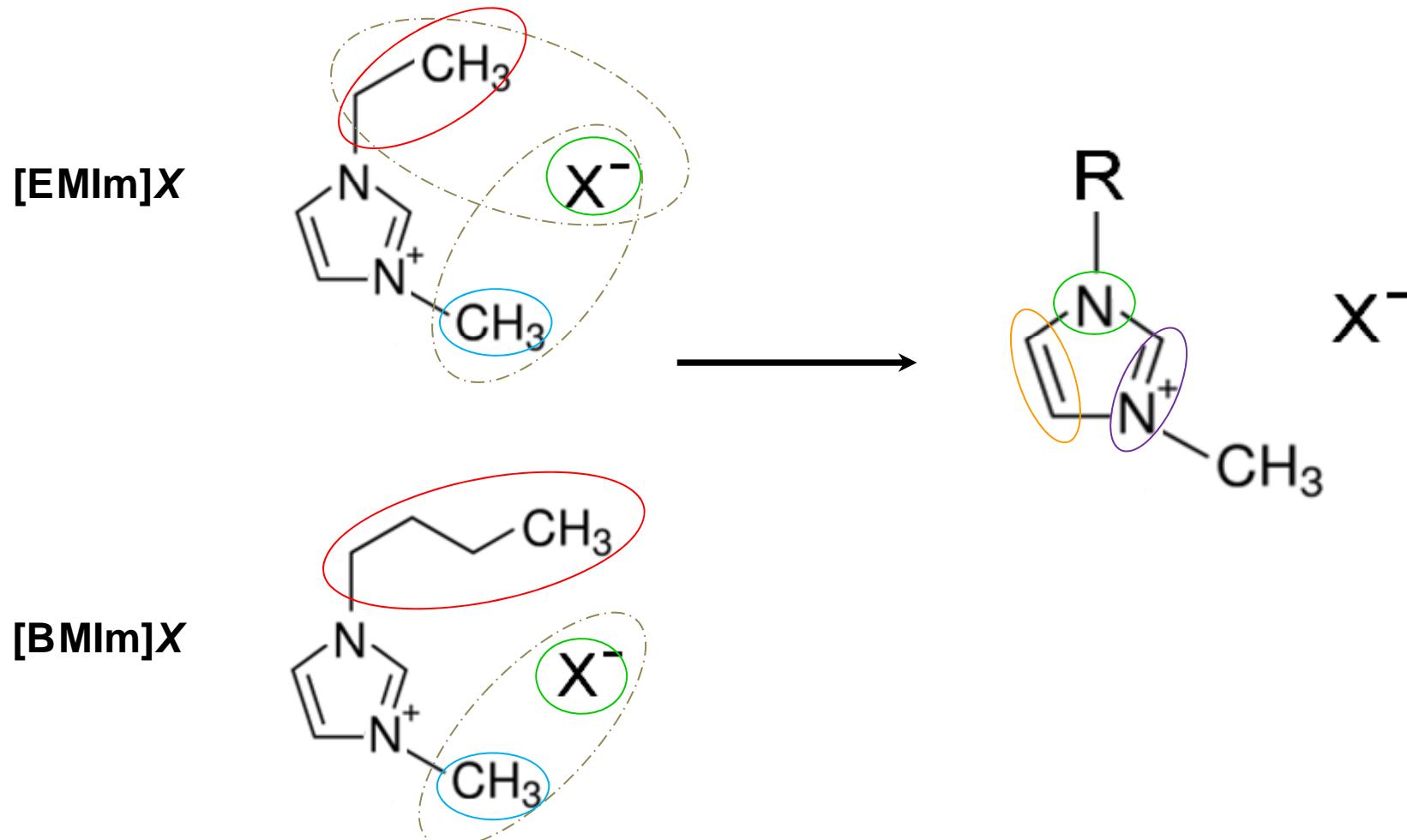


$[EMIm]X$



THERMAL DECOMPOSITION MECHANISM

=> Chain length



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L. Pfützner, C. Ruckhaber, BTU C-S

DFG Deutsche
Forschungsgemeinschaft



DFG - Priority Program 1708:
Material Synthesis near Room Temperature



European Found for Regional Development
Federal State Brandenburg